

Integration of Thermochemical with ORIGEN using the ORIGEN-API

and its application to molten salt reactors

B.W.N. Fitzpatrick and M.H.A. Piro
University of Ontario Institute of Technology
Faculty of Energy Systems and Nuclear Science

Agenda

- What's Thermochemica?
- How can Thermochemica and ORIGEN together solve nuclear fuel problems?
- Integrating (FORTRAN) Thermochemica into C++ code
- Very basics of the ORIGEN-API, and an example
- ORIGEN-API for molten salt reactors
- Integrating Thermochemica into the ORIGEN-API
- Results for a simple molten salt reactor problem

What's Thermochemica?

and why would I want to use it?

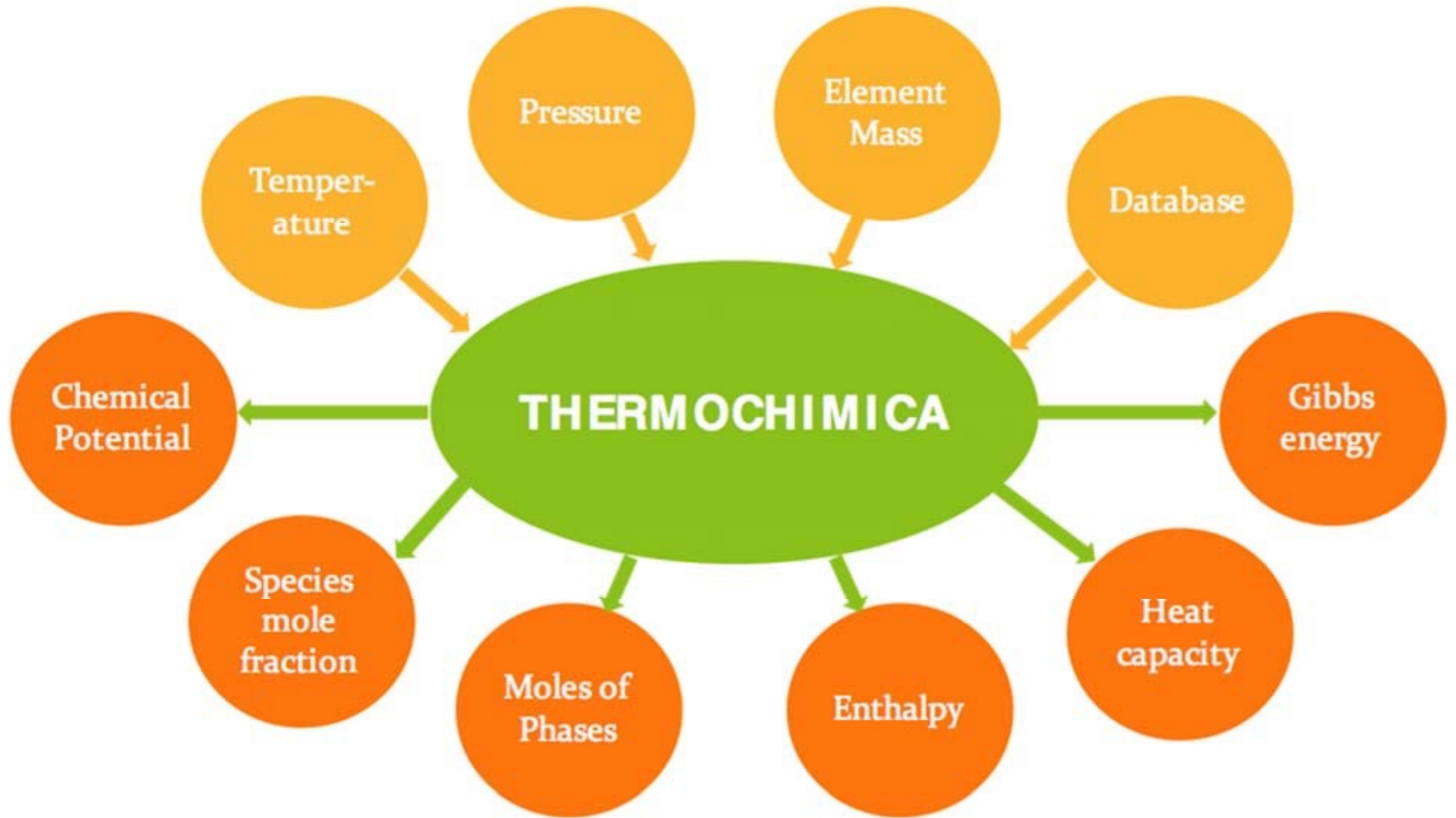
Thermochimica is...

- equilibrium thermodynamics solving system
- particularly good for large, multicomponent, multiphase systems
- designed to work with high-performance multiphysics computing codes
 - open source

Is 'burning' nuclear fuel at equilibrium?

- Close enough
- Chemical equilibrium is attained at relatively short time periods due to the high temperatures of nuclear fuel
- Different elements in nuclear fuel are randomly mixed
- Time scales in nuclear performance simulations are very long (*M.H.A. Piro, J. Banfield, K.T. Clarno, S. Simunovic, T.M. Besmann, B.J. Lewis, et al., Journal of Nuclear Materials. 441 (2013) 240–251*).
- In molten salts, diffusion through the fluid adds to mixing

Input



Output

FactSage / Thermochemica Database File

System U-F-Li

3 2 2 3 12

U F Li

238.02891000 6123456 6123456

gas_ideal

IDMX

LiF

1 1 0.0 1.0 1.0

6000.0000 -351581.57

0.27571767E-07 0.00000000

UF4

1 1 1.0 4.0 0.0

6000.0000 -1639992.8

0.24183333E-06 510660.00

LIQUsoln

SUBG

2.40000 23

Thermochemica and Multiphysics

Thermochemica Output

Physical phenomena

Heat capacity and enthalpy **useful for** **heat transfer calculations**



Speciation of a salt **useful for** **viscosity for momentum equations**



Phase quantities **useful for** **multi-phase flow calculations**



How Thermochemica works...

- Uses the fundamentals of equilibrium thermodynamics to simplify the numerical approach (1st & 2nd laws, minimization of Gibbs free energy)
- Several numerical advantages are obtained, which speed up convergence while increasing numerical stability.
- Read more: *M.H.A. Piro, S. Simunovic, T.M. Besmann, B.J. Lewis, W.T. Thompson, Computational Materials Science, Computational Materials Science. 67 (2013) 266–272.*

How Thermochemica Works

program thermo

```
USE ModuleThermoIO
USE ModuleThermo
USE ModuleGEMSolver
```

implicit none

! Specify units:

```
cInputUnitTemperature = 'K'
cInputUnitPressure   = 'atm'
cInputUnitMass       = 'moles'
cThermoFileName      = '../data/deleteme2.dat'
```

! Specify values:

```
dTemperature      = 553.15D0
dPressure         = 1D0
dElementMass      = 0D0
dElementMass(40)  = 89.24D0  ! Zr
dElementMass(50)  = 0.7D0    ! Sn
dElementMass(8)   = 0.73D0   ! O
dElementMass(1)   = 8.27D0   ! H
dElementMass(41)  = 0.9D0    ! Nb
dElementMass(23)  = 0.16d0   ! V
```

...

How Thermochemica Works

...

! Specify output and debug modes:

iPrintResultsMode = 2

IDebugMode = .FALSE.

!IDebugMode = .TRUE.

! Parse the ChemSage data-file:

call ParseCSDataFile(cThermoFileName)

! Call Thermochemica:

if (INFOThermo == 0) call Thermochemica

! Perform post-processing of results:

if (iPrintResultsMode > 0) call PrintResults

! Destruct everything:

if (INFOThermo == 0) call ResetThermoAll

! Call the debugger:

call ThermoDebug

end program thermo

91.000 mol BCC_A2

```
{ 2.9960E-04 NB:H
+ 2.6446E-05 NB:O
+ 9.5641E-03 NB:VA
+ 2.3302E-04 SN:H
+ 2.0569E-05 SN:O
+ 7.4387E-03 SN:VA
+ 5.3262E-05 V:H
+ 4.7015E-06 V:O
+ 1.7003E-03 V:VA
+ 2.9707E-02 ZR:H
+ 2.6223E-03 ZR:O
+ 0.94833  ZR:VA }
```

Thermochemica Output...

Sublattice 1; stoichiometric coefficient: 1.0000

```
{ Nb      9.8901E-03
+ Sn      7.6923E-03
+ V       1.7582E-03
+ Zr      0.98066  }
```

Sublattice 2; stoichiometric coefficient: 3.0000

```
{ H       3.0293E-02
+ O       2.6740E-03
+ VA      0.96703  }
```

=====
| System properties |
=====

Temperature = 553.15 [K]

Pressure = 1.0000 [atm]

System Component Mass [mol] Chemical potential [J/mol]

Sn 7.0000E-01 -2.601398E+05

Nb 9.0000E-01 -2.424836E+04

Zr 8.9240E+01 -2.166661E+04

V 1.6000E-01 -2.195394E+04

O 7.3000E-01 -5.873069E+05

H 8.2700E+00 -8.083377E+04

Integral Gibbs energy = -3.23819E+06 [J]

Functional norm = 1.25761E-12 [unitless]

of stable pure condensed phases = 0

of stable solution phases = 1

=====
Thermochemica Output ct'd

Recent Developments

Modified Quasi-chemical Model (MQM)

- Of specific interest to molten salts
- Does not focus on chemical species on a lattice, but rather mixing of species in pairs.
- This captures short-range order in liquid or solid solutions

Relevant to molten salt calculations!

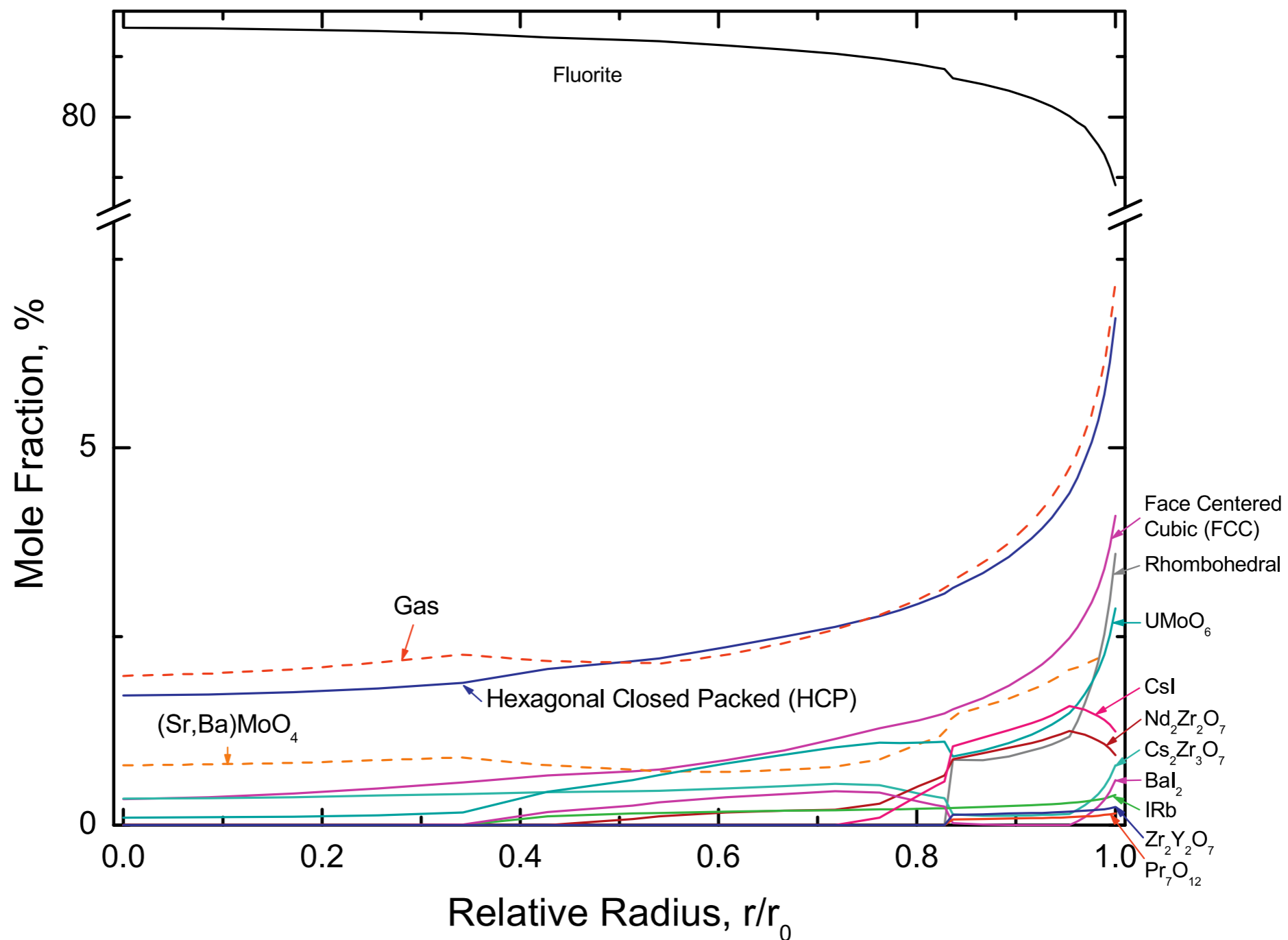
Previous Work

- Work by *M.H.A. Piro* et. al. simulated the irradiation of a nuclear fuel pellet in an LWR
- Combined Thermochemica, ORIGEN and AMP

Previous Work

Origen simulation of fission product formation in solid fuel pellet in an LWR

M.H.A. Piro, J. Banfield, K.T. Clarno, S. Simunovic, T.M. Besmann, B.J. Lewis, et al.,
Journal of Nuclear Materials,
441 (2013) 240–251.

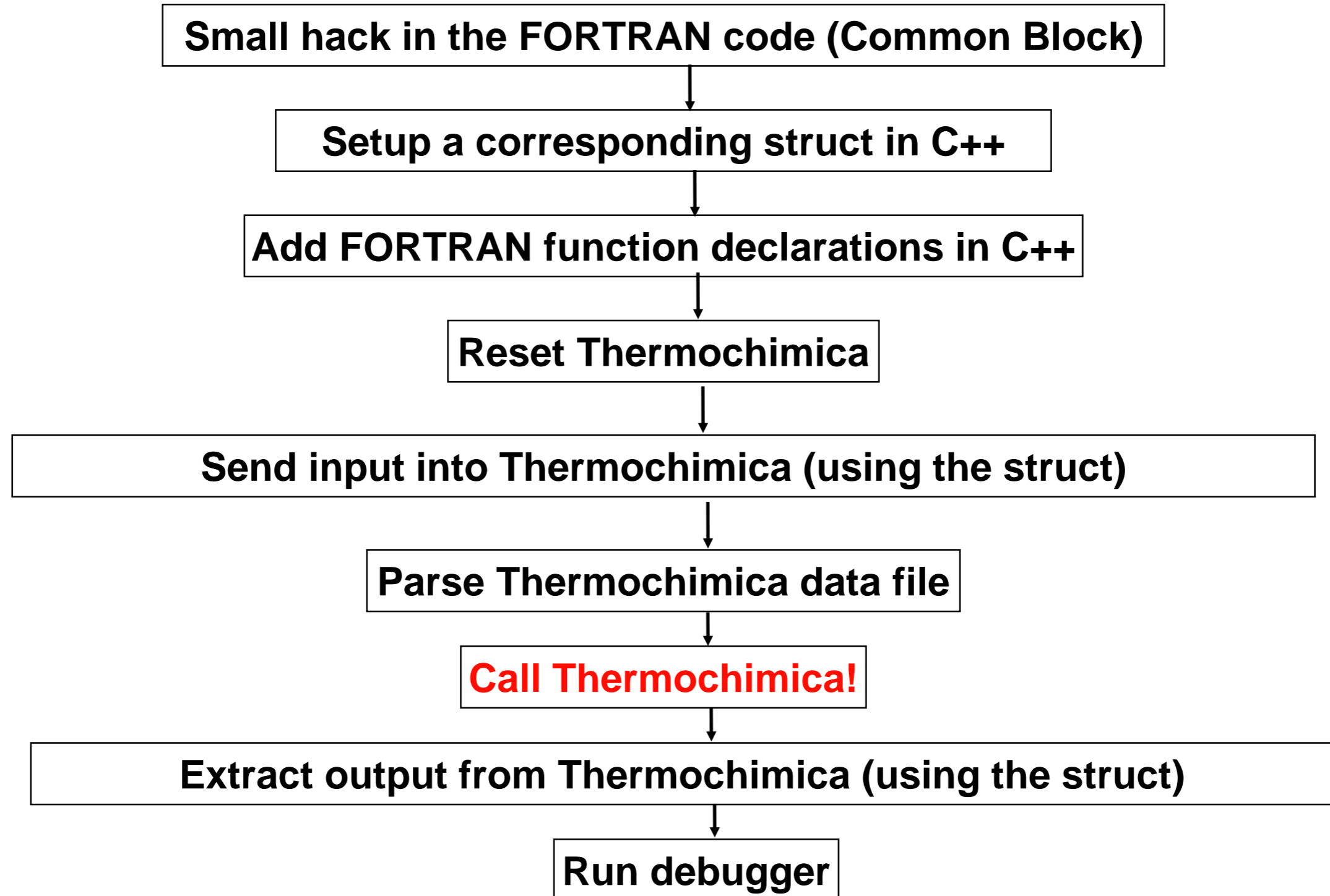


Previous Work

Simulated distribution of phases in solid fuel pellet in an LWR

M.H.A. Piro, J. Banfield, K.T. Clarno, S. Simunovic, T.M. Besmann, B.J. Lewis, et al., *Journal of Nuclear Materials*, 441 (2013) 240–251.

Integrating Thermochemica into C++ Code



Integrating Thermochemica into C++ Code

small hack for input/output...

in `/thermochemica/src/shared/ModuleThermoIO.f90`, we have

```
module ModuleThermoIO
```

```
! INPUT VARIABLES:
```

```
integer          :: iCounter, iPrintResultsMode  
real(8)         :: dTemperature, dPressure  
real(8), dimension(0:118) :: dElementMass  
character(15)    :: cInputUnitTemperature, cInputUnitPressure, cInputUnitMass  
character(120)   :: cThermoFileName
```

```
! OUTPUT VARIABLES
```

```
integer          :: nSolnPhasesOut, nPureConPhaseOut, nSpeciesOut, INFOThermo  
real(8)         :: dGibbsEnergySys  
real(8),dimension(:,allocatable) :: dSolnPhaseMolesOut, dPureConPhaseMolesOut, dSpeciesMoleFractionOut  
character(25),dimension(:,allocatable) :: cSolnPhaseNameOut, cPureConPhaseNameOut, cSpeciesNameOut, cSpeciesPhaseOut  
logical, dimension(:,allocatable) :: lSpeciesStable
```

just add...

`COMMON/THERM/` dTemperature, dPressure, dGibbsEnergySys, dElementMass, iCounter, &
iPrintResultsMode, INFOThermo, nSolnPhasesOut, nPureConPhaseOut, nSpeciesOut, &
cInputUnitTemperature, cInputUnitPressure, cInputUnitMass, cThermoFileName

Integrating Thermochemica into C++ Code

```
extern "C" {
```

```
struct{  
    double dTemperature;  
    double dPressure;  
    double dGibbsEnergySys;  
    double dElementMass[119];  
    int iCounter;  
    int iPrintResultsMode;  
    int INFOThermo;  
    int nSolnPhasesOut;  
    int nPureConPhaseOut;  
    int nSpeciesOut;  
    char cInputUnitTemperature[15];  
    char cInputUnitPressure[15];  
    char cInputUnitMass[15];  
    char cThermoFileName[120];  
} therm_;
```

**corresponding
struct in C++**

Integrating Thermochemica into C++ Code

```
void fortfunc_(int *ii, float *ff);  
void helloworld_(int *ii);  
void parsecsfile0_(char cc[120]);  
void goodbyeworld_(int *ii);  
void parsecsdatafile_(char cc[120]);  
void stringconverter_(char cc[15]);  
void thermochemica_();  
void thermodebug_(); //was int *ii  
void resetthermoall_();  
void printresults_();  
void variablepointers_();  
void resultstofile_();
```

**Add FORTRAN
function
declarations
to C++**

Integrating Thermochemica into C++ Code

**Reset
Thermochemica**

```
resetthermoall_();
```

Integrating Thermochemica into C++ Code

```
therm_.dElementMass[1] = 0.5;  
therm_.dElementMass[8] = 1;  
  
therm_.dTemperature = 553.15;  
therm_.dPressure = 0.5;
```

**Send input to
Thermochemica
(using the
struct)**

Integrating Thermochemica into C++ Code

Access output

```
printf("Species out: %i  nSolnPhasesOut %i, nPureConPhaseOut: %i, Gibbs  
energy:%e  n",therm_.nSpeciesOut,therm_.nSolnPhasesOut,  
therm_.nPureConPhaseOut, therm_.dGibbsEnergySys);
```


Integrating Thermochemica into C++ Code

**Run the
debugger**

```
thermodebug_();
```

ORIGEN API

What's the ORIGEN API?

- Application Programming Interface is software that allows 2 applications to communicate with each other
- The ORIGEN-API is a set of classes for Oak Ridge Isotope GENERation (ORIGEN) for performing depletion / decay calculations in your own software.

(W. A. Wieselquist et al., ORIGEN API v0.5.2 Github page)

An Example

exNEAMS.cpp is an example that ships with the ORIGEN API.

exNEAMS.cpp - Test2

an Introductory Example

Other files to include

Create a library

Extract a transition matrix from it

Initialize a material

Create vectors for time and flux/power

Solve the decay problem for the material

Look at the results!

exNEAMS.cpp

the code

```
#include <algorithm>
#include <iostream>
#include <string>
#include <vector>
```

```
#include "Nemesis/gtest/nemesis_gtest.hh"
#include "Nemesis/harness/DBC.hh"
#include "Origen/Core/dc/FakeFactory.h"
#include "Origen/Core/dc/TransitionMatrixP.h"
#include "Origen/Core/io/LibraryIO.h"
#include "Origen/Core/xf/MultiZoneDepleter.h"
#include "Origen/Core/xf/Solver_Fake.h"
#include "Origen/Manager/libld/TransitionMatrixUpdater.h"
#include "Origen/Solver/matrex/Solver_matrex.h"
#include "ScaleUtils/IO/DB.h"
#include "Standard/Interface/AbstractWriter.h"
#include "Standard/Interface/BasicIOWriter.h"
#include "Standard/Interface/Communicator.h"
#include "Standard/Interface/jdebug.h"
```

```
using ScaleUtils::IO::nprintf;
using namespace Origen;
```

```
typedef TransitionMatrixP TransitionMatrix;
typedef SP_TransitionMatrixP SP_TransitionMatrix;
typedef std::vector<double> Vec_Dbl;
typedef std::shared_ptr<Vec_Dbl> SP_Vec_Dbl;
```

```
// GLOBAL COMMUNICATOR
Standard::Communicator world;
```

**other files to
include**

exNEAMS.cpp

the code

Creating a library

```
// Get a general 2237-nuclide ORIGEN library.  
SP_Library lib( new Library() );  
FakeFactory::Library_scale_pwr( *lib );
```

exNEAMS.cpp

the code

Extracting a transition matrix

```
// Extract a transition matrix from it. We will use this PWR transition  
// matrix for all materials in this test.  
SP_TransitionMatrix trx( lib->newsp_transition_matrix_at( 0 ) );
```


exNEAMS.cpp

the code

Initializing a material

```
// Create a single 3.6% enriched UO2 material.
SP_Material mat;
{
    std::vector<int> ids;           // nuclide ids in IZZZAAA format
    std::vector<double> numden;    // atoms/barn-cm
    double volume = 5.6;          // cm^3
    int id = 1234;                 // material id
    std::string name = "material_1234"; // material name

    // Use the FakeFactory to get a reasonable initial composition.
    FakeFactory::vera_uox_e360( ids, numden );

    // Initialize the material.
    mat = SP_Material( new Material( lib, name, id, volume ) );

    // Set the beginning of step number densities.
    mat->set_numden_bos( numden, ids );
}
```

exNEAMS.cpp

the code

**Create vectors
for time and
flux/power**

```
// Create depletion times/fluxes.  
// 0 --> 3 days at 1e14 n/cm^2s  
// 3 --> 250 days at 1e14 n/cm^2s  
// 250 --> 500 days at 1e14 n/cm^2s  
// 500 --> 515 days at 0 (decay)  
// 515 --> 530 days at 0 (decay)  
std::vector<double> time{0, 3, 250, 500, 515, 530};  
std::vector<double> flux{1, 1, 1, 0, 0};  
for( size_t j = 0; j < time.size(); ++j )  
    time[j] *= 86400.0; // scale to seconds  
for( size_t j = 0; j < flux.size(); ++j )  
    flux[j] *= 1e14; // scale to n/cm^2s  
size_t nsteps = flux.size();
```

exNEAMS.cpp

the code

```
// Create a Bell depletion/decay solver.  
Solver_matrex slv;
```

```
// Enter step loop.
```

```
for( size_t j = 0; j < nsteps; ++j )
```

```
{
```

```
    // Add a new time step to this material.
```

```
    double dt = time[j + 1] - time[j];
```

```
    mat->add_step( dt );
```

```
    mat->set_flux( flux[j] );
```

```
    mat->set_transition_matrix( trx );
```

```
    // Solve the step using only data on the material, solve takes
```

```
    // the bos vector, time, flux, and pointer to eos.
```

```
    SP_Vec_Dbl n0 = mat->amount_bos();
```

```
    SP_Vec_Dbl n1 = mat->amount_eos();
```

```
    slv.set_transition_matrix( &*mat->transition_matrix() );
```

```
    slv.solve( *n0, mat->flux(), mat->dt(), &*n1 );
```

```
    slv.clear();
```

```
}
```

**Solve the decay
problem**

exNEAMS.cpp

the code

**Look at the
results!**

```
// Inspect results stored in material.  
if( false ) std::cout << mat->to_string() << std::endl;
```

Let's make this more applicable to molten salt reactors

Let's do a fictive simulation of the MSRE,
by modifying `exNEAMS.cpp`.

exNEAMS.cpp - Modifying for Molten Salt Simulation

Other files to include

Create a library

Extract a transition matrix from it

Initialize a material

Create vectors for time and flux/power

Solve the decay problem for the material

Look at the results!

exNEAMS.cpp - Modifying for Molten Salt Simulation

Other files to include

Create a library

Extract a transition matrix from it

Initialize a material

fuel-bearing
molten salt

Create vectors for time and flux/power

Solve the decay problem for the material

Look at the results!

exNEAMS.cpp - Modifying for Molten Salt Simulation

Other files to include

Create a library

Extract a transition matrix from it

Initialize a material

fuel-bearing
molten salt

Create vectors for time and flux/power

Solve the decay problem for the material

MSRE
power
history

Look at the results!

exNEAMS.cpp

Modifying for Molten Salt Simulation

**Fuel-bearing
molten salt**

```
//FakeFactory::vera_uox_e360( ids, numden );
```

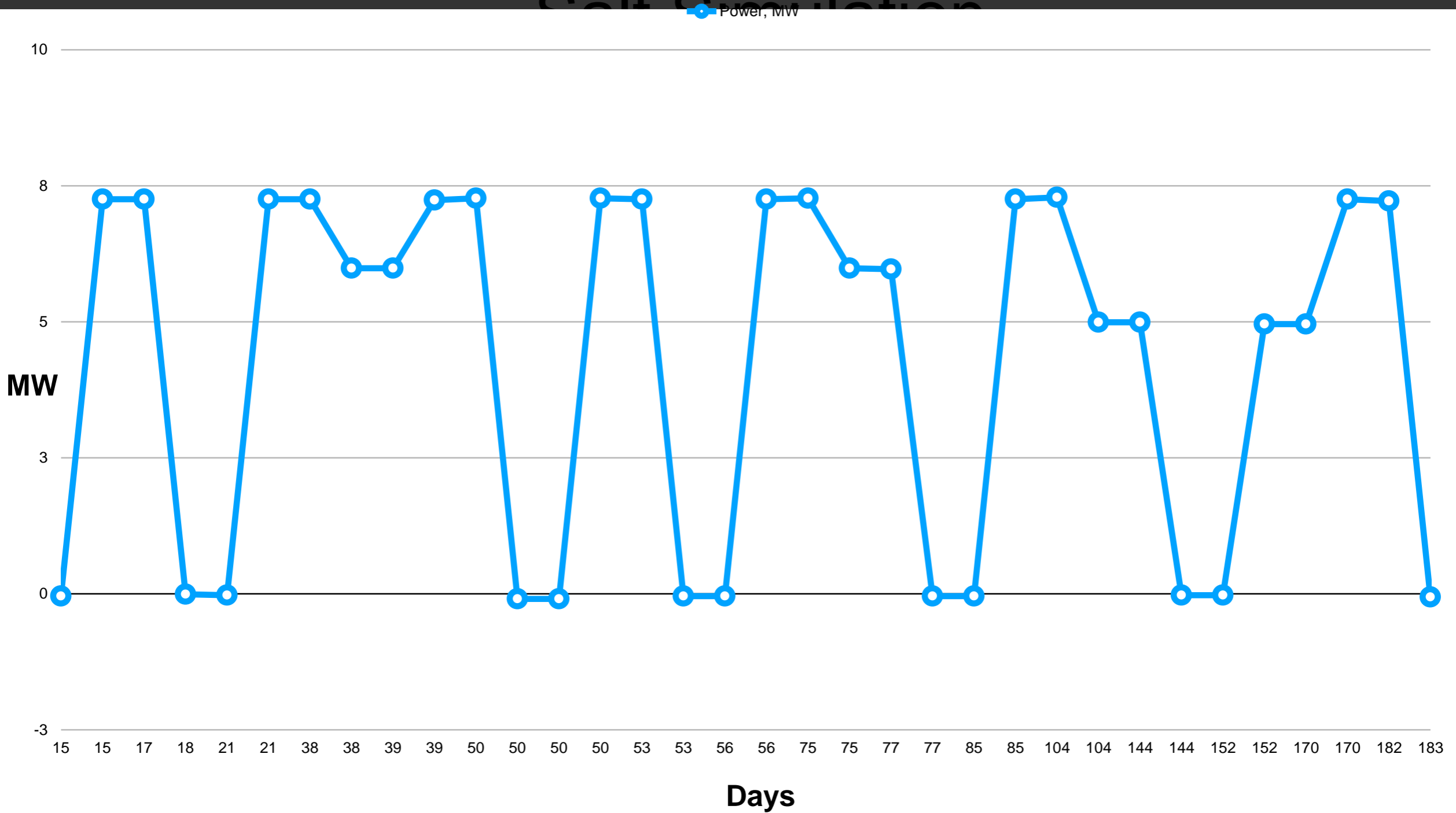
```
std::map<int, double> map_numden;  
map_numden[3007] = 4.13510E-02; // Li-7  
map_numden[4009] = 1.23417E-02; // Be-9  
map_numden[9019] = 7.20399E-02; // F-19  
map_numden[40090] = 6.54746E-04; // Zr-90  
map_numden[40091] = 1.42502E-04; // Zr-91  
map_numden[40092] = 2.18206E-04; // Zr-92  
map_numden[40094] = 2.21259E-04; // Zr-94  
map_numden[40096] = 3.56255E-05; // Zr-96  
map_numden[92234] = 1.14510E-08; // U-234  
map_numden[92235] = 8.01574E-05; // U-235  
map_numden[92238] = 1.48852E-04; // U-238  
ScaleSTL::map_to_vectors( map_numden, &ids, &numden );
```

${}^7\text{LiF}-\text{BeF}_2-\text{ZrF}_4-\text{UF}_4$
(65-29-5-1 mole %)

35% enriched ${}^{235}\text{U}$

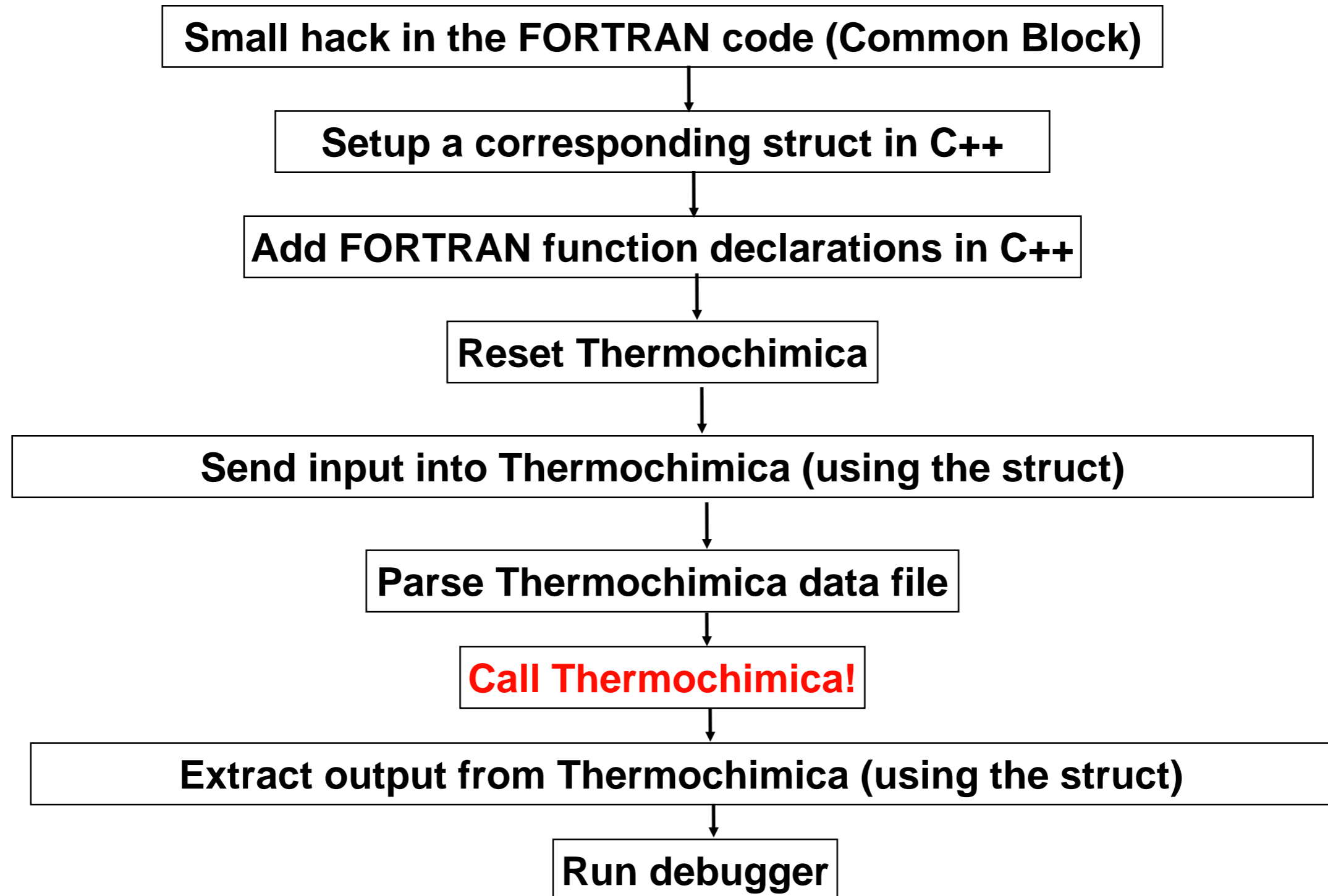
exNEAMS.cpp

Modifying for Molten Salt Simulation

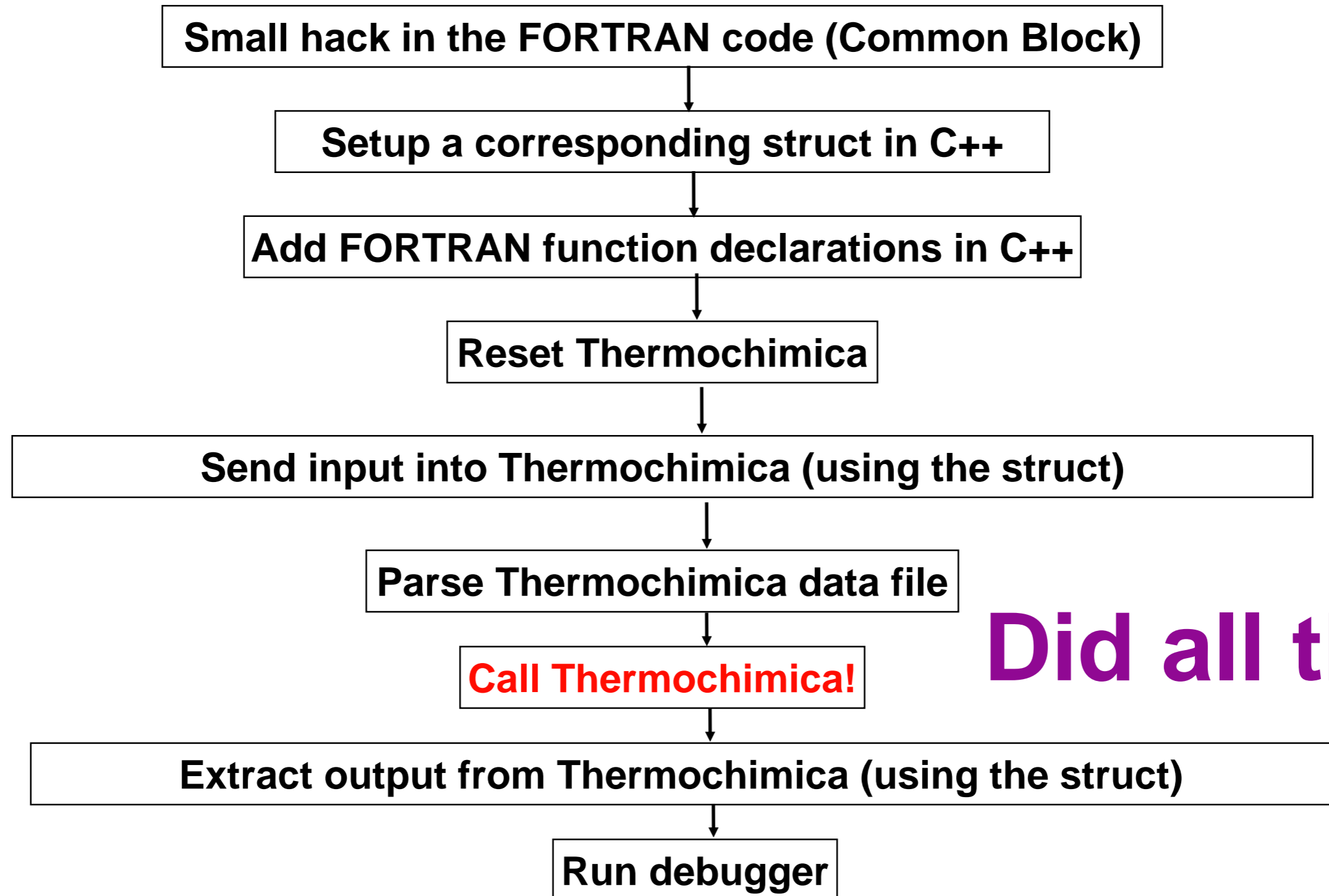


Let's bring in the thermochemistry
to our mini-MSRE experiment.

Integrating Thermochemica into C++ Code



Integrating Thermochemica into C++ Code



Did all that.

exNEAMS.cpp - Test2 an Introductory Example

Other files to include

Create a library

Extract a transition matrix from it

Initialize a material

Create vectors for time and flux/power

Adjust temperature based on power

Solve the decay problem for the material

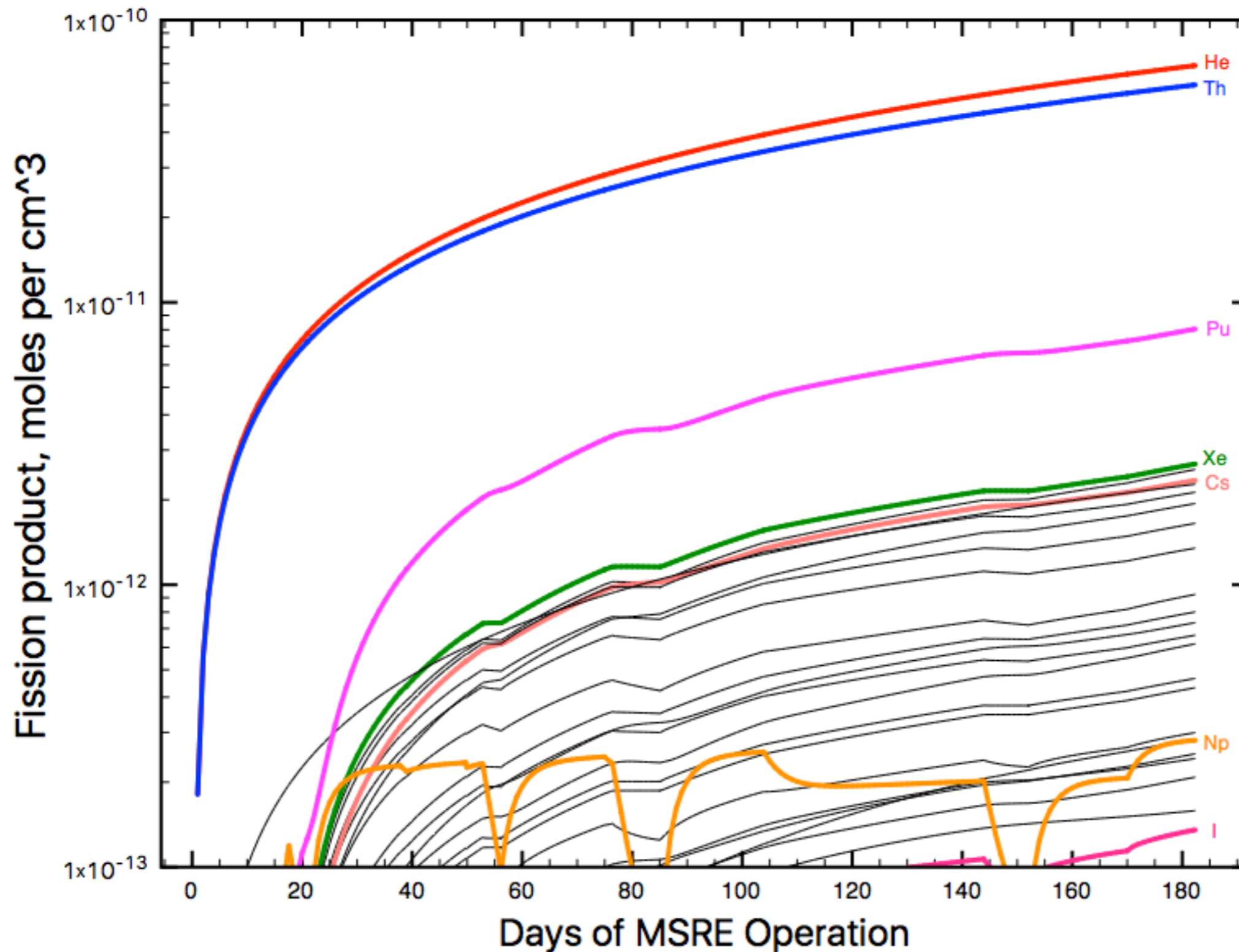
Combine isotopes into elements
for Thermochemica

Run Thermochemica

Look at the results!

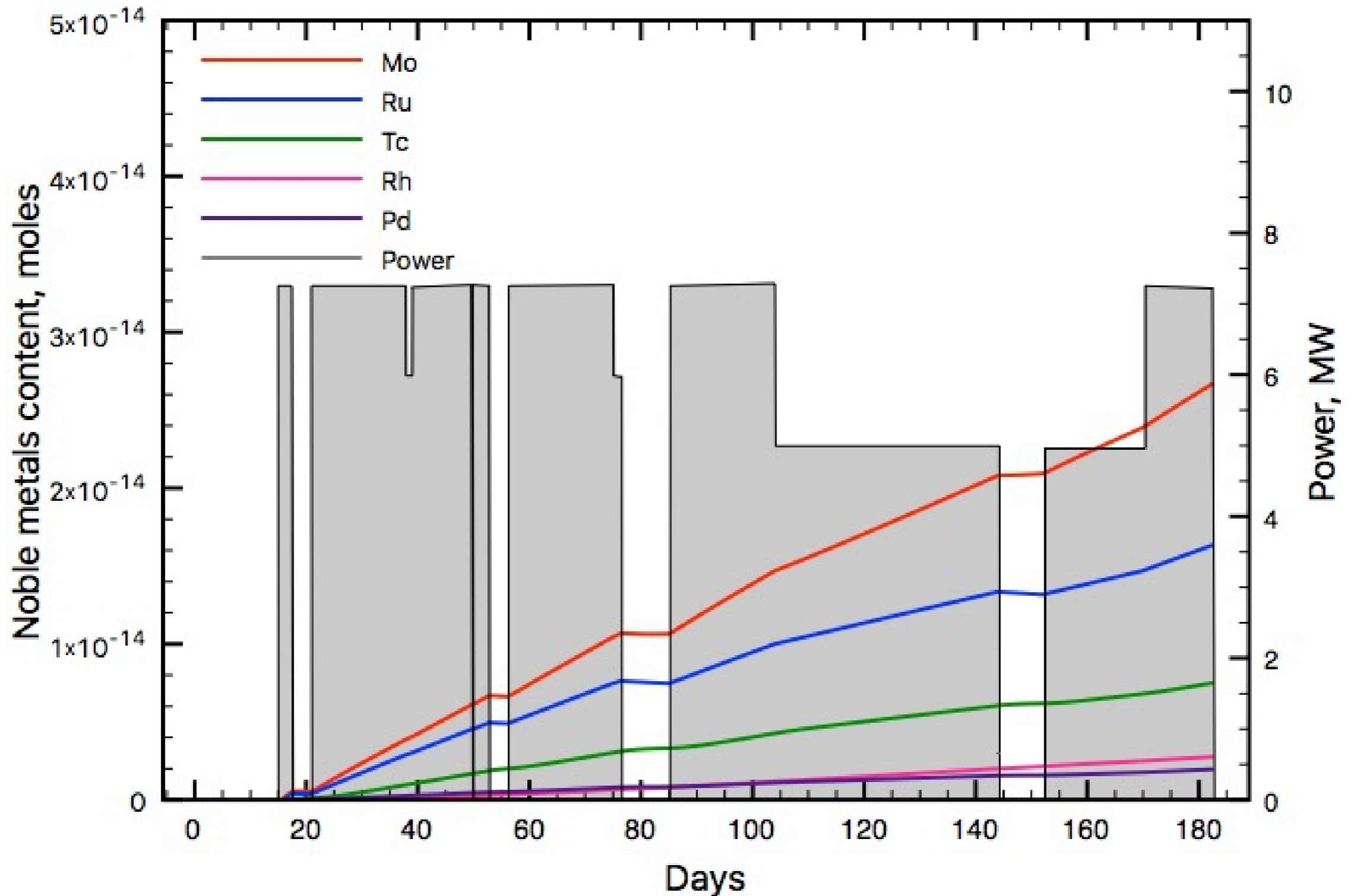
Results

Fission Product Formation During Second Campaign of MSRE



Results

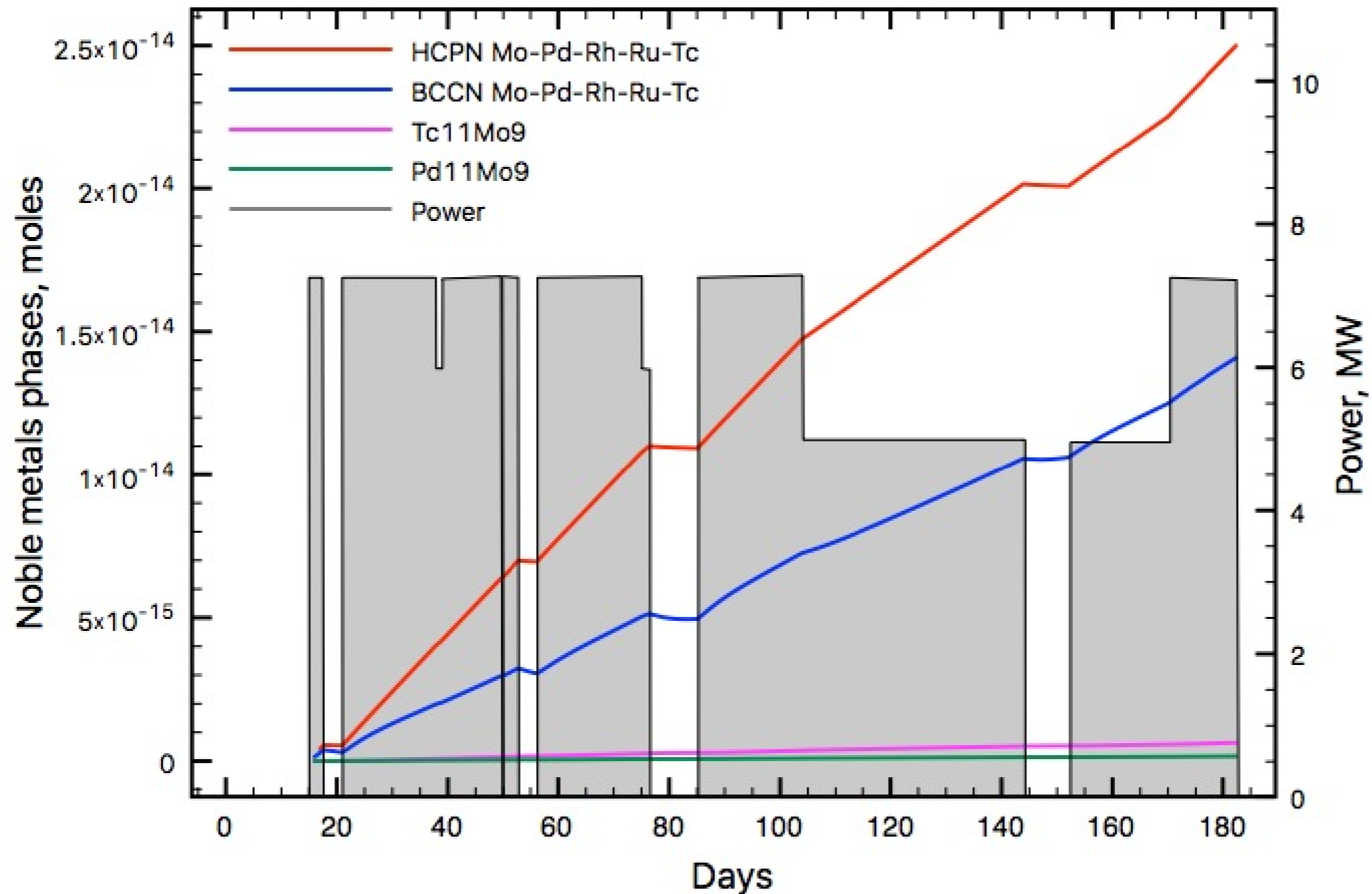
Noble metals formation from ORIGEN-API, for MSRE



Power history taken from: M. Rosenthal, R. Briggs, P. Haubenreich, al. Molten salt reactor program semiannual progress report for period ending august 31, ORNL-4622. USA: Oak Ridge National Laboratory, 1970: 38-41, (1968)

Results

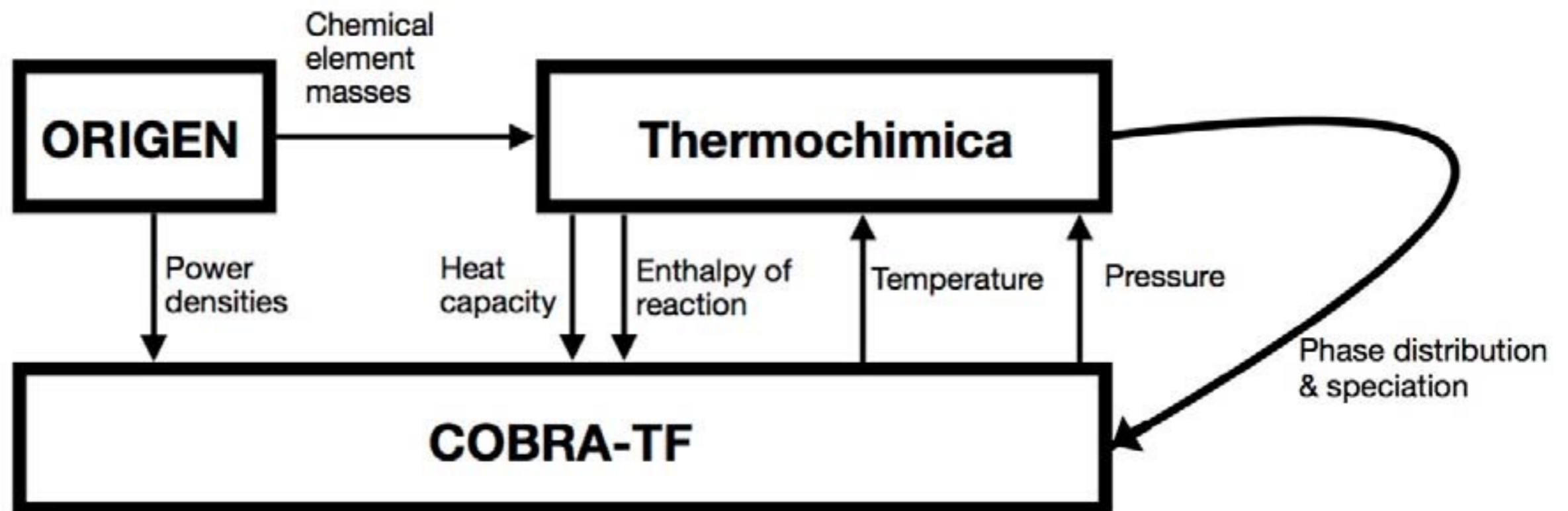
Phase evolution of noble metals formation from ORIGEN-API / Thermochemica coupling, for MSRE



Power history taken from: M. Rosenthal, R. Briggs, P. Haubenreich, al. Molten salt reactor program semiannual progress report for period ending august 31, ORNL-4622. USA: Oak Ridge National Laboratory, 1970: 38-41, (1968)

Future Work

Incorporating **thermal-hydraulics** for temperature, pressure, and transport



B.W.N. Fitzpatrick et al., Proceedings of the ANS Annual Meeting, Philadelphia, 2018.

Future Work

Expanding the thermodynamics database through experimental work



Netzsch Jupiter STA 449 F1: DSC + TGA for thermodynamic measurements on salts



Glovebox for fabricating salts

Future Work

Viscometer?

Expanding the thermodynamics database through experimental work



Netzsch Jupiter STA 449 F1: DSC + TGA for thermodynamic measurements on salts



Glovebox for fabricating salts

Acknowledgements

- This research was undertaken, in part, thanks to the funding from the Canada Research Chairs (950-231328) and Discovery Grant programs of the Natural Sciences and Engineering Research Council of Canada.
- Special thanks to Ben Collins, Robert Salko, Robert Taylor, Jake McMurray, Ted Besmann, Z. Taylor for all the useful discussions